Exam III Solutions Administered: Friday, November 5, 2021 16 points

For each problem part: 0 points if not attempted or no work shown, 1 point for partial credit, if work is shown, 2 points for correct numerical value of solution, if work is shown

Problem 1. (8 points)

For an ideal mixture, the volume of the mixture, V_{mix} , is given by the sum of the pure component molar volumes, V_i , weighted by the mole fraction, x_i . Similarly, the enthalpy of the mixture, H_{mix} , is given by the sum of the pure component enthalpies, H_i , weighted by the mole fraction. As a reminder, the sum of the mole fractions is unity.

$$V_{mix} = \sum_{i=1}^{n_c} x_i V_i \qquad H_{mix} = \sum_{i=1}^{n_c} x_i H_i \qquad 1 = \sum_{i=1}^{n_c} x_i$$

where n_c is the number of components. Now consider a three component ideal mixture ($n_c = 3$) with the following pure component properties and mixture properties.

component	А	В	С	mixture
molar volume (liter/mol)	19	13	11	15.0
enthalpy (kJ/mol)	42	65	51	52.0

(a) Is this system of algebraic equations linear or nonlinear? (2 pts)

(b) Determine the composition of this mixture. Show reasoning and method. (6 pts)

Solution:

This is a set of linear algebraic equations, which can be written as follows.

$$x_A V_A + x_B V_B + x_C V_C = V_{mix}$$

$$x_A H_A + x_B H_B + x_C H_C = H_{mix}$$

$$x_A + x_B + x_C = 1$$

This set of linear algebraic equations can be written in matrix form as

 $\underline{Ax} = \underline{b}$

where
$$\underline{A} = \begin{bmatrix} V_A & V_B & V_C \\ H_A & H_B & H_C \\ 1 & 1 & 1 \end{bmatrix}$$
, $\underline{b} = \begin{bmatrix} V_{mix} \\ H_{mix} \\ 1 \end{bmatrix}$ and $\underline{x} = \begin{bmatrix} x_A \\ x_B \\ x_C \end{bmatrix}$.

To solve this problem, I wrote the following Matlab script, xm03p01 f21.m.

clear all; A = [19.0 13.0 11.0 42.0 65.0 51.0 1.0 1.0 1.0]; b = [15.0 52.0 1.0]; detA = det(A) invA = inv(A) x = invA*b

At the command line prompt, I executed the script with the command

>> xm03p01_f21

which generated the following output

Therefore the composition of the mixture is given by

	$[x_A]$		[0.4154]	
$\underline{x} =$	x_B	=	0.3385	
	$\begin{bmatrix} x_C \end{bmatrix}$		0.2462	

Problem 2. (8 points)

Consider a non-ideal mixture in which the enthalpy of the mixture is given by the expression

$$H_{mix} = \sum_{i=1}^{n_c} \sum_{j \ge i}^{n_c} (2 - \delta_{ij}) \Omega_{ij} x_i x_j$$

where δ_{ij} is the Kronecker delta function and is 1 for i = j and 0 for $i \neq j$. For a three-component mixture, this equation becomes

$$H_{mix} = \Omega_{AA} x_A x_A + 2\Omega_{AB} x_A x_B + 2\Omega_{AC} x_A x_C + \Omega_{BB} x_B x_B + 2\Omega_{BC} x_B x_C + \Omega_{CC} x_C x_C x_C$$

The expressions for the molar volume and sum of the mole fractions remain unchanged from problem 1. The mixing parameters, $\Omega_{ij} = \sqrt{H_i H_j}$, where the pure component enthalpies are given in problem 1. The same numerical values of the mixture properties are observed as those in the problem 1.

(a) Is this system of algebraic equations linear or nonlinear? (2 pts)

(b) Determine the composition of this mixture. Show reasoning and method. (6 pts)

Solution:

This is a set of non-linear algebraic equations, which can be written as follows.

$$f_{1}(x_{A}, x_{B}, x_{C}, x_{D}) = x_{A}V_{A} + x_{B}V_{B} + x_{C}V_{C} - V_{mix}$$

$$f_{2}(x_{A}, x_{B}, x_{C}, x_{D}) = x_{A} + x_{B} + x_{C} - 1$$

$$f_{3}(x_{A}, x_{B}, x_{C}, x_{D}) = \Omega_{AA}x_{A}x_{A} + 2\Omega_{AB}x_{A}x_{B} + 2\Omega_{AC}x_{A}x_{C}$$

$$\stackrel{\rightarrow}{\leftarrow} \stackrel{\rightarrow}{\leftarrow} \stackrel{\leftarrow}{\leftarrow} \stackrel{\leftarrow}{\leftarrow} \qquad + \Omega_{BB}x_{B}x_{B} + 2\Omega_{BC}x_{B}x_{C}$$

$$+ \Omega_{CC}x_{C}x_{C} - H_{mix}$$

I will solve this using the Newton Raphson Method with Numerical Approximations to the Derivatives, as implemented in the code nrndn.m.

This code requires that I input my system of nonlinear algebraic equations in the function, funkeval.m.

```
function f = funkeval(x)
  these two lines force a column vector of length n
00
%
n = max(size(x));
f = zeros(n, 1);
8
% enter the functions here
%
VA = 19.0;
VB = 13.0;
VC = 11.0;
Vmix = 15.0;
HA = 42.0;
HB = 65.0;
HC = 51.0;
omega AA = sqrt(HA*HA);
```

```
omega_AB = sqrt(HA*HB);
omega_AC = sqrt(HA*HC);
omega_BB = sqrt(HB*HB);
omega_BC = sqrt(HB*HC);
omega_CC = sqrt(HC*HC);
Hmix = 52.0;
f(1) = VA*x(1) + VB*x(2) + VC*x(3) - Vmix;
f(2) = x(1) + x(2) + x(3) - 1.0;
f(3) = omega_AA*x(1)*x(1) + 2*omega_AB*x(1)*x(2) + 2*omega_AC*x(1)*x(3) ...
+ omega_BB*x(2)*x(2) + 2*omega_BC*x(2)*x(3) ...
+ omega_CC*x(3)*x(3) - Hmix;
```

The Newton Raphson method requires an initial guess. I will use the solution from problem 1 as my initial guess. I want the tolerance to be 1.0^{-6} . I set the print flag to 1. At the command line prompt, I executed the following commands:

clear all; x0 = [0.4154; 0.3385; 0.2462]; tol = 1.0e-6; iprint = 1; [x,err,f] = nrndn(x0,tol,iprint)

This command provided the following output.

Because the error is less than the specified tolerance, the Newton Raphson method has converged. Therefore the composition of the mixture is given by

$$\underline{x} = \begin{bmatrix} x_A \\ x_B \\ x_C \end{bmatrix} = \begin{bmatrix} 0.4079 \\ 0.3683 \\ 0.2237 \end{bmatrix}$$

This solution is not very sensitive to the initial guess. All of the following initial guesses converged to this solution.

x0 = [1.0/3.0; 1.0/3.0; 1.0/3.0]; x0 = [0.4154; 0.3385; 0.2462]; x0 = [0.5; 0.25; 0.25]; x0 = [0.25; 0.5; 0.25]; x0 = [0.25; 0.25; 0.5];