

Exam III Solutions
Administered: Wednesday, November 10, 2017
16 points

For each problem part: 0 points if not attempted or no work shown,
1 point for partial credit, if work is shown,
2 points for correct numerical value of solution, if work is shown

Problem 1. (8 points)

For an ideal mixture, the volume of the mixture, V_{mix} , is given by the sum of the pure component molar volumes, V_i , weighted by the mole fraction, x_i . Similarly, the enthalpy of the mixture, H_{mix} , is given by the sum of the pure component enthalpies, H_i , weighted by the mole fraction. The cost of the mixture, C_{mix} , is also given by the sum of the pure component costs, C_i , weighted by the mole fraction. As a reminder, the sum of the mole fractions is unity.

$$V_{mix} = \sum_{i=1}^{n_c} x_i V_i \quad H_{mix} = \sum_{i=1}^{n_c} x_i H_i \quad C_{mix} = \sum_{i=1}^{n_c} x_i C_i \quad 1 = \sum_{i=1}^{n_c} x_i$$

where n_c is the number of components. Now consider a four component ideal mixture ($n_c = 4$) with the following pure component properties and mixture properties.

component	A	B	C	D	mixture
molar volume (liter/mol)	11	14	17	8	12.8
enthalpy (kJ/mol)	49	74	45	63	57.1
cost (\$/mol)	40.18	89.67	38.22	55.99	56.216

- (a) Is this system of algebraic equations linear or nonlinear? (2 pts)
(b) Determine the composition of this mixture. Show reasoning and method. (6 pts)

Solution:

This is a set of linear algebraic equations, which can be written as follows.

$$\begin{aligned} x_A V_A + x_B V_B + x_C V_C + x_D V_D &= V_{mix} \\ x_A H_A + x_B H_B + x_C H_C + x_D H_D &= H_{mix} \\ x_A C_A + x_B C_B + x_C C_C + x_D C_D &= C_{mix} \\ x_A + x_B + x_C + x_D &= 1 \end{aligned}$$

This set of linear algebraic equations can be written in matrix form as

$$\underline{Ax} = \underline{b}$$

where $\underline{\underline{A}} = \begin{bmatrix} V_A & V_B & V_C & V_D \\ H_A & H_B & H_C & H_D \\ C_A & C_B & C_C & C_D \\ 1 & 1 & 1 & 1 \end{bmatrix}$, $\underline{\underline{b}} = \begin{bmatrix} V_{mix} \\ H_{mix} \\ C_{mix} \\ 1 \end{bmatrix}$ and $\underline{\underline{x}} = \begin{bmatrix} x_A \\ x_B \\ x_C \\ x_D \end{bmatrix}$.

To solve this problem, I wrote the following Matlab script, xm03p01_f17.m.

```
clear all;
A = [11      14      17      8
     49      74      45      63
     40.18   89.67   38.22   55.99
     1       1       1       1];
b = [12.8
     57.1
     56.216
     1.0];
detA = det(A)
invA = inv(A)
x = inv(A)*b
```

At the command line prompt, I executed the script with the command

```
>> xm03p01_f17
```

which generated the following output

```
detA =  -1.2314e+03

invA =
    -0.3336    -0.3327     0.1681    14.2197
    -0.0291    -0.0723     0.0585     1.5112
     0.2417     0.1591    -0.0950    -6.6363
     0.1210     0.2459    -0.1316    -8.0946

x =
    0.4000
    0.3000
    0.2000
    0.1000
```

Therefore the composition of the mixture is given by

$$\underline{\underline{x}} = \begin{bmatrix} x_A \\ x_B \\ x_C \\ x_D \end{bmatrix} = \begin{bmatrix} 0.4 \\ 0.3 \\ 0.2 \\ 0.1 \end{bmatrix}$$

Problem 2. (8 points)

Consider a non-ideal mixture in which the enthalpy of the mixture is given by the expression

$$H_{mix} = \sum_{i=1}^{n_c} \sum_{j \geq i}^{n_c} (2 - \delta_{ij}) \Omega_{ij} x_i x_j$$

where δ_{ij} is the Kronecker delta function and is 1 for $i = j$ and 0 for $i \neq j$. For a four-component mixture, this equation becomes

$$\begin{aligned} H_{mix} = & \Omega_{AA} x_A x_A + 2\Omega_{AB} x_A x_B + 2\Omega_{AC} x_A x_C + 2\Omega_{AD} x_A x_D \\ & + \Omega_{BB} x_B x_B + 2\Omega_{BC} x_B x_C + 2\Omega_{BD} x_B x_D \\ & + \Omega_{CC} x_C x_C + 2\Omega_{CD} x_C x_D \\ & + \Omega_{DD} x_D x_D \end{aligned}$$

The expressions for the molar volume, cost and sum of the mole fractions remain unchanged from problem 1. The mixing parameters, $\Omega_{ij} = \sqrt{H_i H_j}$, where the pure component enthalpies are given in problem 1. The same numerical values of the mixture properties are observed as those in the problem 1.

- (a) Is this system of algebraic equations linear or nonlinear? (2 pts)
 (b) Determine the composition of this mixture. Show reasoning and method. (6 pts)

Solution:

This is a set of non-linear algebraic equations, which can be written as follows.

$$\begin{aligned} f_1(x_A, x_B, x_C, x_D) &= x_A V_A + x_B V_B + x_C V_C + x_D V_D - V_{mix} \\ f_2(x_A, x_B, x_C, x_D) &= x_A C_A + x_B C_B + x_C C_C + x_D C_D - C_{mix} \\ f_3(x_A, x_B, x_C, x_D) &= x_A + x_B + x_C + x_D - 1 \\ f_4(x_A, x_B, x_C, x_D) &= \Omega_{AA} x_A x_A + \Omega_{AB} x_A x_B + \Omega_{AC} x_A x_C + \Omega_{AD} x_A x_D \\ &\quad + \Omega_{BB} x_B x_B + \Omega_{BC} x_B x_C + \Omega_{BD} x_B x_D \\ &\quad + \Omega_{CC} x_C x_C + \Omega_{CD} x_C x_D + \Omega_{DD} x_D x_D - H_{mix} \end{aligned}$$

I will solve this using the Newton Raphson Method with Numerical Approximations to the Derivatives, as implemented in the code `nrndn.m`.

This code requires that I input my system of nonlinear algebraic equations in the function, `funkeval.m`.

```
function f = funkeval(x)
%
%   these two lines force a column vector of length n
%
n = max(size(x));
f = zeros(n,1);
%
%   enter the functions here
%
```

```

VA = 11;
VB = 14;
VC = 17;
VD = 8;
Vmix = 12.8;
CA = 40.18;
CB = 89.67;
CC = 38.22;
CD = 55.99;
Cmix = 56.216;
HA = 49;
HB = 74;
HC = 45;
HD = 63;
omega_AA = sqrt(HA*HA);
omega_AB = sqrt(HA*HB);
omega_AC = sqrt(HA*HC);
omega_AD = sqrt(HA*HD);
omega_BB = sqrt(HB*HB);
omega_BC = sqrt(HB*HC);
omega_BD = sqrt(HB*HD);
omega_CC = sqrt(HC*HC);
omega_CD = sqrt(HC*HD);
omega_DD = sqrt(HD*HD);
Hmix = 57.1;
f(1) = VA*x(1) + VB*x(2) + VC*x(3) + VD*x(4) - Vmix;
f(2) = CA*x(1) + CB*x(2) + CC*x(3) + CD*x(4) - Cmix;
f(3) = x(1) + x(2) + x(3) + x(4) - 1.0;
f(4) = omega_AA*x(1)*x(1) + 2*omega_AB*x(1)*x(2) + 2*omega_AC*x(1)*x(3) + 2*omega_AD*x(1)*x(4) ...
      + omega_BB*x(2)*x(2) + 2*omega_BC*x(2)*x(3) + 2*omega_BD*x(2)*x(4) ...
      + omega_CC*x(3)*x(3) + 2*omega_CD*x(3)*x(4) ...
      + omega_DD*x(4)*x(4) - Hmix;

```

The Newton Raphson method requires an initial guess. I will use the solution from problem 1 as my initial guess. I want the tolerance to be 1.0^{-6} . I set the print flag to 1. At the command line prompt, I executed the following commands:

```

>> x0 = [0.4; 0.3; 0.2; 0.1];
>> tol = 1.0e-6;
>> iprint = 1;
>> [x,err,f] = nrndn(x0,tol,iprint)

```

This command provided the following output.

```

iter =    1, err = 1.33e-01 f = 3.04e-01
iter =    2, err = 3.57e-04 f = 8.20e-04
iter =    3, err = 2.56e-09 f = 5.89e-09

x =
    0.2032
    0.2573
    0.2941
    0.2455

err =    2.5631e-09

f =    5.8889e-09

```

Because the error is less than the specified tolerance, the Newton Raphson method has converged. Therefore the composition of the mixture is given by

$$\underline{x} = \begin{bmatrix} x_A \\ x_B \\ x_C \\ x_D \end{bmatrix} = \begin{bmatrix} 0.2032 \\ 0.2573 \\ 0.2941 \\ 0.2455 \end{bmatrix}$$

This solution is not very sensitive to the initial guess. All of the following initial guesses converged to this solution.

```
x0 = [0.4; 0.3; 0.2; 0.1];
x0 = [0.25; 0.25; 0.25; 0.25];
x0 = [0.7; 0.1; 0.1; 0.1];
x0 = [0.1; 0.7; 0.1; 0.1];
x0 = [0.1; 0.1; 0.7; 0.1];
x0 = [0.1; 0.1; 0.1; 0.7];
```