

Multiscale Materials Modeling

Lecture 06

Polymers



These notes created by David Keffer, University of Tennessee, Knoxville, 2012.



Multiscale Modeling of Polymers

- I. Introduction
- II. Atomistic Simulation
- III. Coarse-Grain Potential Generation
- IV. Coarse-Grain Simulation
- V. Validation
- VI. More Examples
- VII. Conclusions

Motivation



The dynamic phenomena of polymers span many orders of magnitude.

The shortest relaxation time, the vibration of individual chemical bonds in the polymer take place on a subfemtosecond (10⁻¹⁵ s) time scale.

The longest relaxation, corresponding to the relaxation of the end-to-end vector of the polymer chain is much larger and is a strong function of polymer identity, degree of linearity, temperature, and chain length.



The KWW model, a stretched exponential, is frequently use to describe the longest relaxation mode in a polymer.

$$X(t) = \exp\left(-\left(\frac{t}{\tau}\right)^{\beta}\right)$$

As an example, the relaxation time of PET at 1 atm and 563 K has a relaxation time of

DP	simulation method	ρ (g/cm ³)	$\frac{D}{(10^{-10} \text{m}^2/\text{s})}$	η (10 ⁻² Pa·s)	$\begin{array}{c} \langle R_{ete} \rangle \\ (\rm \dot{A}) \end{array}$	$\langle R_g \rangle$ (A)	τ _{ΚWW} (ns)	
		1.0	0.186	5.38	1.0	1.0	7.5	
4	atomistic MD	1.29 ± 0.01	0.40 ± 0.09	0.65 ± 0.07		8.9 ± 5.2	7.5 5.6	
	CGMD-scaled	1.22 ± 0.03	0.50 ± 0.02	0.54 ± 0.06	17.5 ± 6.2	8.1 ± 1.7	5.5	
6	atomistic MD	1.29 ± 0.01	0.17 ± 0.02	1.95 ± 0.65	26.8 ± 10.2	11.2 ± 4.4	15.8	
	CGMD-scaled	1.29 ± 0.01	0.14 ± 0.01	1.40 ± 0.48	22.1 ± 7.8	9.48 ± 5.4	17.4	
8	atomistic MD	1.29 ± 0.01	0.10 ± 0.03	2.23 ± 0.60	28.6 ± 11.2	12.5 ± 5.3	25.3	
	CGMD-scaled	1.29 ± 0.01	0.09 ± 0.01	2.21 ± 0.48	24.7 ± 9.1	10.49 ± 6.3	26.8	
10	atomistic MD	1.29 ± 0.01	0.07 ± 0.01	3.03 ± 0.80	34.2 ± 9.4	13.2 ± 3.8	38.6	
	CGMD-scaled	1.29 ± 0.01	0.06 ± 0.01	2.31 ± 0.16	28.3 ± 10.2	12.18 ± 6.4	59.1	
20	CGMD-scaled	1.18 ± 0.02	0.030 ± 0.006	2.37 ± 0.16	44.2 ± 15.9	19.6 ± 7.0	257.9	
30	CGMD-scaled	1.20 ± 0.01	0.015 ± 0.004	3.50 ± 0.70	56.6 ± 21.0	24.7 ± 9.1	712.9	
40	CGMD-scaled	1.25 ± 0.02	0.008 ± 0.004	8.07 ± 0.91	63.0 ± 23.0	28.0 ± 7.7	3684.0	
50	CGMD-scaled		0.005 ± 0.002	13.37 ± 2.2	69.8 ± 22.7	30.9 ± 10.6	7551.0	
ta madie recimology								

Wang et al., Macromolecules, 2010



This relaxation time scales nonlinearly with chain length.

$$\tau = a(DP)^{\beta}$$

As an example, the relaxation time of PET at 1 atm and 563 scales as

Table 4. Scaling Exponents for Various Properties As a Function of Chain Length and Degree of Model Resolution

DP	simulation method	D	η	τ_{KWW}	$\langle R_{ete} \rangle$	$\langle R_g \rangle$
1-10	atomistic MD	-2.01	0.96	2.78	0.594	0.571
4-10	atomistic MD	-1.91	1.6	2.81	0.59	0.57
20-50	CGMD	-2.0	2.0	3.7	0.51	0.50
Rouse model	N/A	-1	1	2	0.59	0.59
reptation model	N/A	-2	3	3	0.50	0.50

Motivation



relaxation time of PET at 1 atm and 563 as a function of chain length

degree of polymerization	t (ns)	t (s)		
20	258	0.000000258		
50	7551	0.00000755		
100	92072.22	9.21E-05		
500	35507300	0.036		
1000	4.61E+08	0.461		
5000	1.78E+11	178		
10000	2.31E+12	2313		

Clearly, long-chains are out of the reach of molecular simulation.



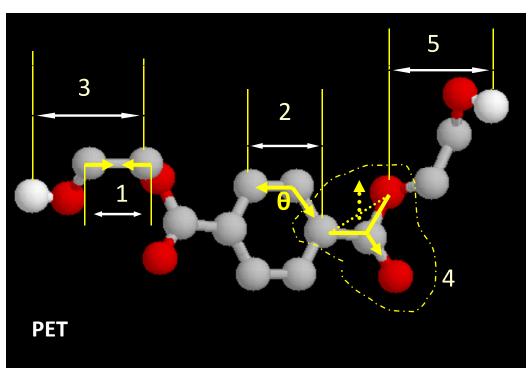
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Length and time scale in polymeric systems:



- \triangleright simulation of long chain polymer is in order of μ s
- → dt = 0.2 fs for stretching interactions in molecular model
- From fs (10⁻¹⁵ s) to μ s (10⁻⁶ s): 9 orders!



- 1: bond stretching
- 2: bond bending
- 3: bond torsion
- 4: out of plane bending
- 5: intramolecular L-J and electrostatic potential (for atoms over four bonds)

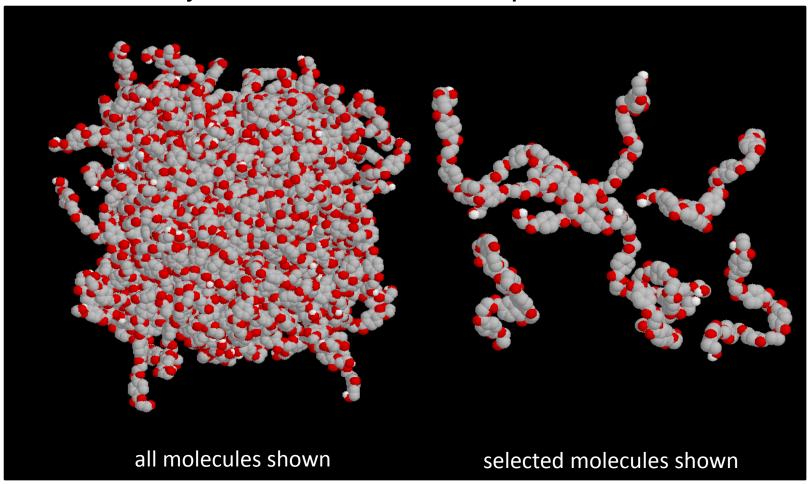
Note: not all pairs are listed O: red; C: grey; H: white

Hedenqvist *et al*. Macromolecules 1998. (neglected OH interactions) OH interactions from Bin Chen et al. J. Phys. Chem. B, 2001

PET Properties



Molecular Dynamic Simulation Snapshots



Snapshots of PET hexamer at T = 563 K and p = 0.13 kPa.

PET Properties: Structural Information



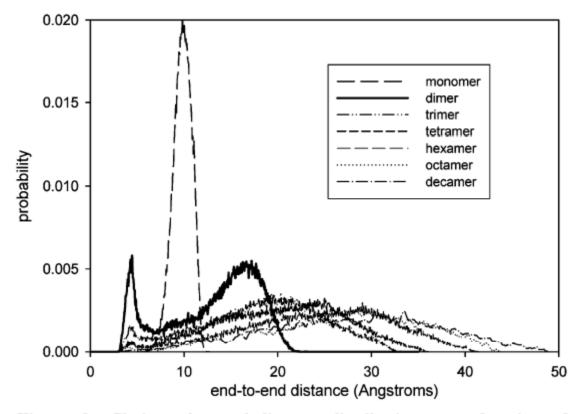


Figure 3. Chain end-to-end distance distributions as a function of degree of polymerization (DP).

PET Properties: Dynamic Information



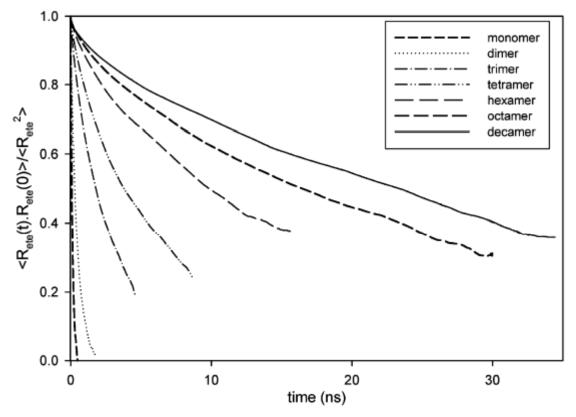


Figure 5. Chain end-to-end distance (R_{ete}) autocorrelation functions as a function of observation time for all DPs studied.

PET Properties: Thermodynamic Information



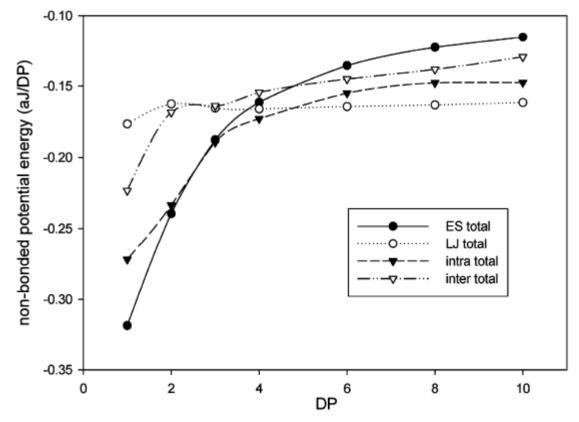


Figure 8. Individual nonbonded contributions to the potential energy as a function of DP. ES = electrostatic; LJ = Lennard-Jones; intra = intramolecular; inter = intermolecular. These contributions are normalized by the degree of polymerization.

PET Properties: Transport Information



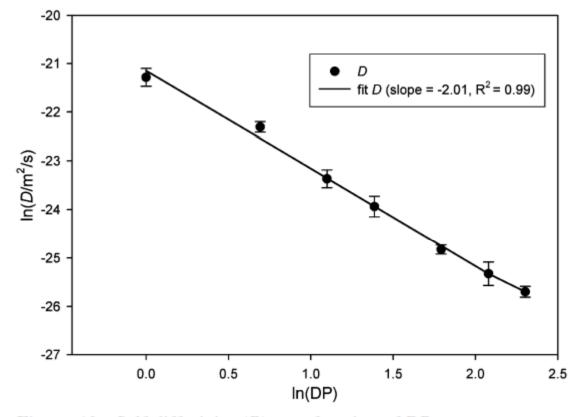


Figure 12. Self-diffusivity (*D*) as a function of DP.

PET Properties



MD Simulation data for PET (DP = 1, 2, 3, 4, 6, 8 and 10) at p = 0.13 kPa, T = 563 K.

DP	1	2	3	4	6	8	10
N	216	216	216	125	125	125	125
V (10 ⁵ Å ³)	0.70 ± 0.04	1.21 ± 0.06	1.71 ± 0.03	1.28 ± 0.01	1.87 ± 0.01	2.46 ± 0.01	3.06 ± 0.01
ρ (g/cm³)	1.30 ± 0.01	1.26 ± 0.01	1.28 ± 0.02	1.29 ± 0.01	1.29 ± 0.01	1.29 ± 0.01	1.29 ± 0.0
U (10 ² aJ)	-0.15 ±0.01	0.20 ± 0.05	1.26± 0.03	1.39± 0.05	3.38 ± 0.17	5.44± 2.03	7.43± 4.06
H (10 ² aJ)	-0.15 ±0.01	0.20 ± 0.05	1.26± 0.03	1.38± 0.05	3.38± 0.17	5.44± 2.03	7.43± 4.06
C_p (10 ³ J/K/kg)	2.73 ± 0.09	2.56 ± 0.06	2.47 ± 0.06	2.45 ± 0.07	2.35 ± 0.05	2.29 ± 0.04	2.25 ± 0.05
β (10-1 GPa-1)	2.17 ± 0.77	3.02 ± 1.12	4.70 ± 3.27	4.51 ± 1.65	3.44 ± 0.44	6.40 ± 3.06	5.21 ± 4.43
α (10 ⁻⁴ K ⁻¹)	6.73 ± 1.02	5.26± 0.38	4.81± 0.55	4.80± 0.17	4.53± 0.27	4.28± 0.16	3.75± 0.79
D (10 ⁻¹⁰ m ² /sec)	5.68 ± 1.14	2.05 ± 0.23	0.71 ± 0.14	0.40 ± 0.09	0.17 ± 0.02	0.10 ± 0.03	0.07 ± 0.01
η (10 ⁻² Pa.s)	0.27 ± 0.01	0.55 ± 0.08	0.58 ± 0.09	0.65 ± 0.07	1.95± 0.65	2.23± 0.60	3.03± 0.80
vs (103 m/sec)	2.01	1.75	1.39	1.41	1.62	1.18	1.31
λ _{GK} (W/m/K)	0.16	0.16	0.13	0.14	0.16	0.12	0.13
λ _B (W/m/K)	0.21 ± 0.02	0.20 ± 0.05	0.31 ± 0.02	0.13 ± 0.11	0.12 ± 0.07	0.22 ± 0.03	0.18 ± 0.01
H-bond (2.0 Å)(%)	54.22	42.72	42.57	38.84	28.55	22.25	20.55
H-bond (2.5 Å)(%)	69.50	57.16	55.11	51.02	38.01	30.18	27.44
<r<sub>ete> (A)</r<sub>	9.8 ± 1.0	13.6 ± 4.9	18.3 ± 6.6	21.1 ± 7.5	26.8 ± 10.2	28.6 ± 11.2	34.2 ± 9.4
<r<sub>a> (Å)</r<sub>	4.0 ± 0.7	5.6 ± 2.8	7.6 ± 2.7	8.9 ± 5.2	11.2 ± 4.4	12.5 ± 5.3	13.2 ± 3.8
R _{SE} (Å)	4.06	5.54	14.96	23.70	19.23	28.02	31.9
TR (ns)	0.128	0.552	2.42	5.59	15.26	27.8	37.5
TKWW (ns)	0.129	0.502	2.81	5.59	15.78	25.3	38.6

Wang, Q., Keffer, D.J., Petrovan, S., Thomas, J.B., in preparation, 2009.



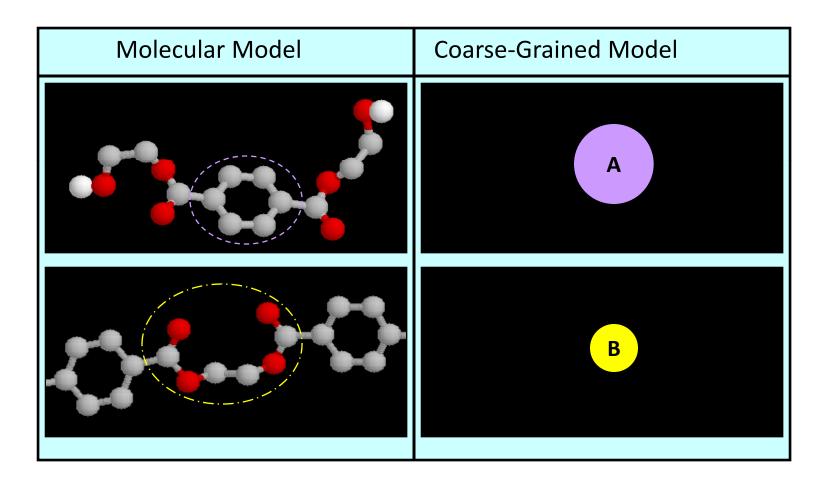
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Polymer Mesoscale Models



Example: PET

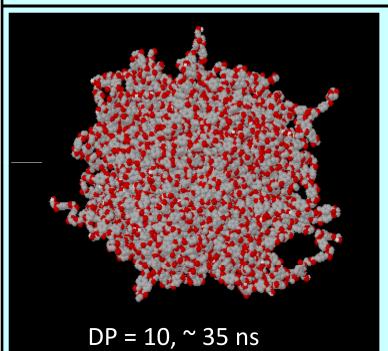


Mptivation: Medesdare M60018e-Grained Simulations



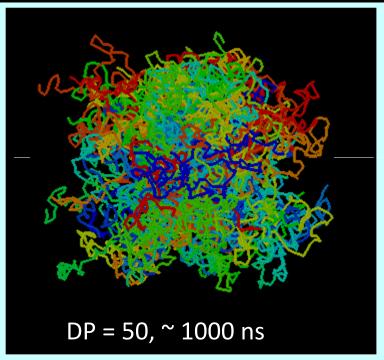
Exampl

Molecular System



- many degrees of freedom
- computationally expensive
- limited to short chains (1-10)

Coarse-Grained System



- fewer degrees of freedom
- computationally modest
- long chains (25-50 or longer)

$$F = ma$$
 $F \equiv -\nabla U$

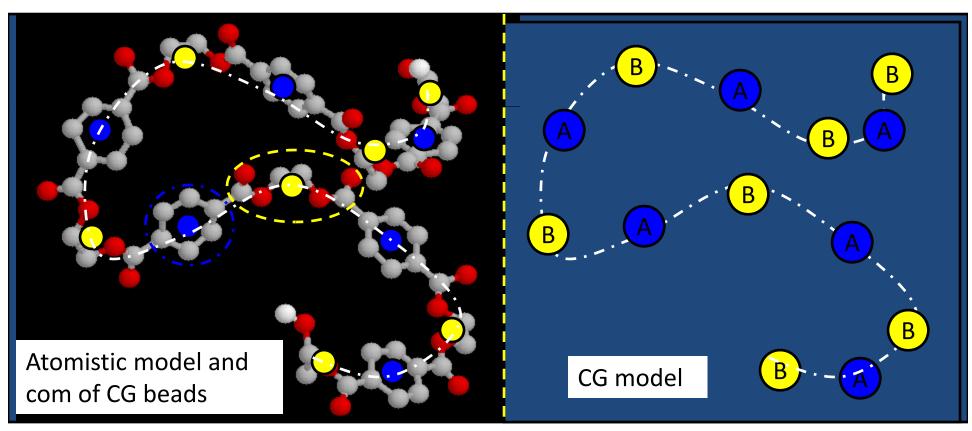
$$\frac{d^2 x_{i,\alpha}}{dt} = -\frac{1}{m} \frac{\partial U}{\partial x}$$

Harmandaris, Adhikari, van der Vegt, Kremer, Macromofecules, 2006.

Coarse-Grained model: based on atomistic MD simulat



Atomistic MD simulation is limited to DP up to 10, CG MD simulation will allow us to get to DP from 25 to 50 or even longer chains by neglecting some local degrees of freedom.



CG model on top of atomistic model for hexamer (DP=6)

Fundamentals of Sustainable Technology

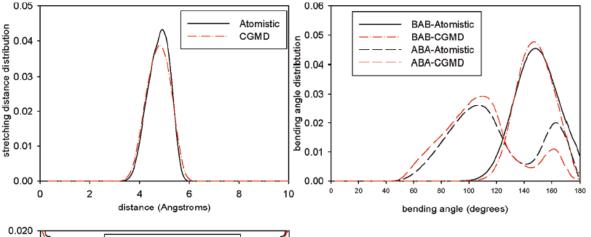
CG chain for hexamer, total interaction centers is 13 for each molecule (original:90).

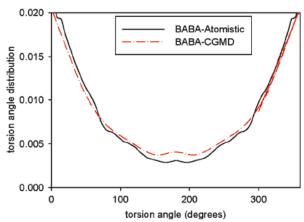


Step 1. Run atomistic MD simulations on short chains.

Generate Pair Correlation functions of center-of-mass of molecular fragments corresponding

to beads.





Stretching, bending and torsion modes indicate distributions of intramolecular degrees of freedom.

This is a representation of the structure of the material.

Remember: PET is an ABABABA polymer.

Coarse-Grained Potentials: parameterization



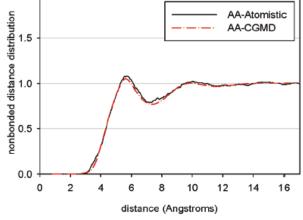
Step 1. Run atomistic MD simulations on short chains.

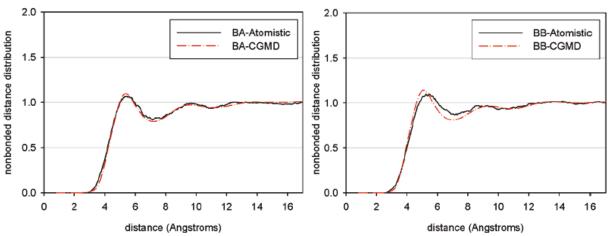
Generate Pair Correlation functions of center-of-mass of molecular fragments corresponding

to beads.

Non-bonded distributions of intermolecular (and distant intramolecular) degrees of freedom.

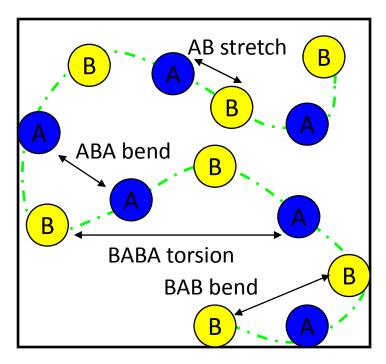
Remember: PET is an ABABABA polymer.







Step 2. From the PCFs obtained from atomistic simulations, estimate the CG interaction potential.



CG chain model and intramolecular interaction for hexamer,

- 1: bond stretching (BA);
- 2: bond bending (BAB and ABA);
- 3: bond torsion (BABA)

$$U^{CG} = U^{CG}_{stretch} + U^{CG}_{bend} + U^{CG}_{torsion} + U^{CG}_{non-bond}$$

$$U^{CG}_{stretch}(r,T) = -k_B T \ln P^{CG}_{stretch}(r,T) + C_r$$

$$U^{CG}_{bend}(\theta,T) = -k_B T \ln P^{CG}_{bend}(\theta,T) + C_{\theta}$$

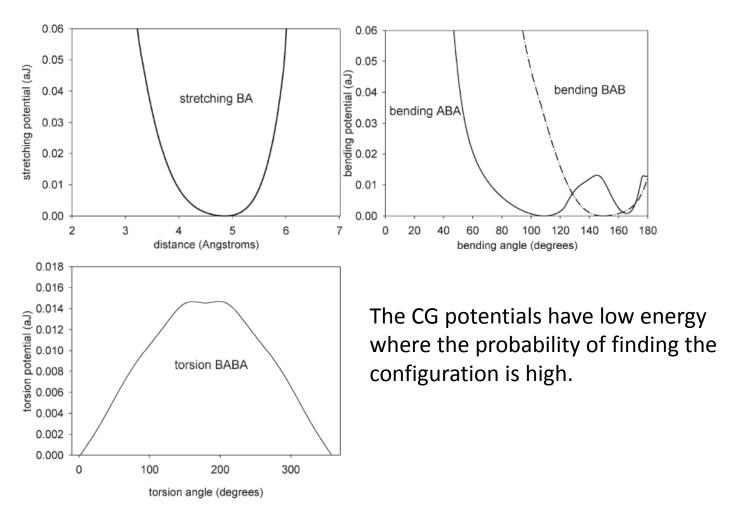
$$U^{CG}_{torsion}(\phi,T) = -k_B T \ln P^{CG}_{torsion}(\phi,T) + C_{\phi}$$

For bonded modes, we use a simple Boltzmann distribution, which assumes that all of the bonded modes are decoupled.

This assumption won't work for non-bonded modes.

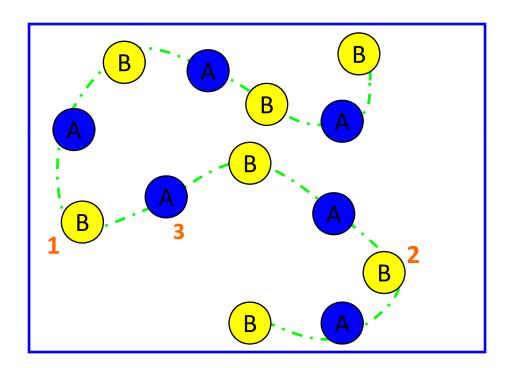


Step 2. From the PCFs obtained from atomistic simulations, estimate the CG interaction potential.





Step 2. From the PCFs obtained from atomistic simulations, estimate the CG interaction potential.



There are a variety of ways to obtain the non-bonded interaction potentials from the PCFs.

One of the most-common is iterative Boltzmann inversion.



Iterative Boltzmann Inversion

IBI: iterative, can reproduce structures from atomistic simulation.¹

$$U_{\alpha\beta,i+1}(r) = U_{\alpha\beta,i}(r) + k_{B}T \ln \left(\frac{g_{\alpha\beta,i}(r)}{g_{\alpha\beta}(r)}\right)$$

$$U_{\alpha\beta,0}(r) = -k_{B}T \ln \left(g_{\alpha\beta}(r)\right)$$

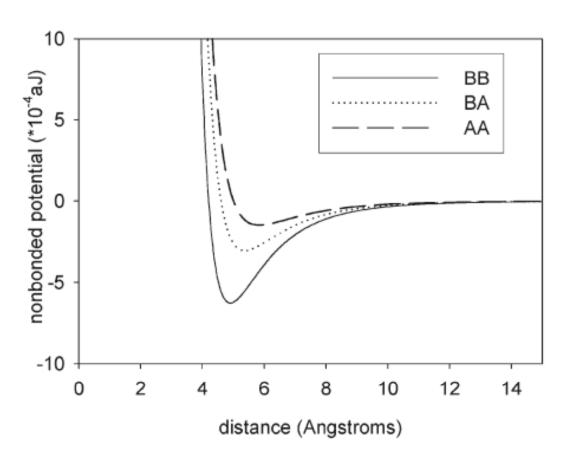
Potential is iteratively refined to reproduce the target PCF.

May involve many CG simulations to obtain the PCFs.

Computationally expensive, may not work for Inhomogeneous multi-component system

1 D. Reith, M. Putz and F. Muller-Plathe, J. Comput. Chem. 24, 1624 (2003)





The IBI method or an alternative procedure leads to a non-bonded interaction potential for the coarse-grained beads.



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Motivation: Molecular vs. Coarse-Grained Simulations



The Coarse Grain Simulations use the same code as the atomistic molecular dynamics simulations because both are solving Newton's equations of motion given an interaction potential.

In the atomistic simulation, the particles are atoms and the interaction potentials are atom-atom potentials.

In the CG simulation, the particles are CG beads and the interaction potentials are bead-bead potentials.

$$F = ma$$
 $F \equiv -\nabla U$ $\frac{d^2 x_{i,\alpha}}{dt} = -\frac{1}{m} \frac{\partial U}{\partial x_{i,\alpha}}$

Before Analyzing results, you must verify that your coarse-grained potential can correctly reproduce the structure of the atomistic system.



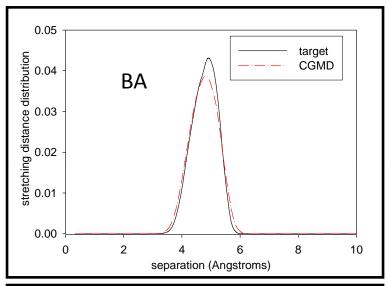
Multiscale Modeling of Polymers

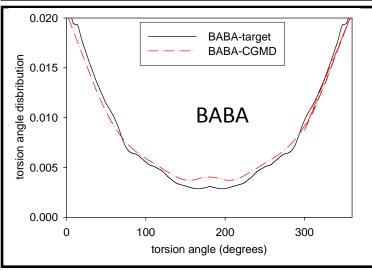
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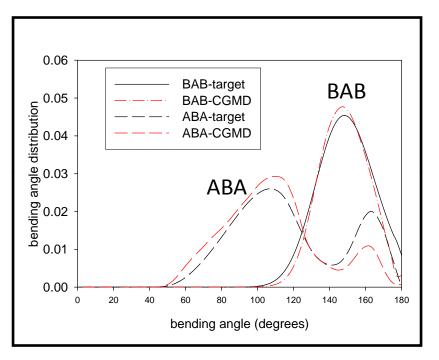
Coarse-Grained MD simulation: validation



We compared the P(r), $P(\theta)$ and $P(\phi)$ from CG and atomistic MD simulation.





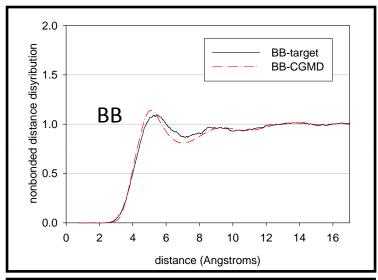


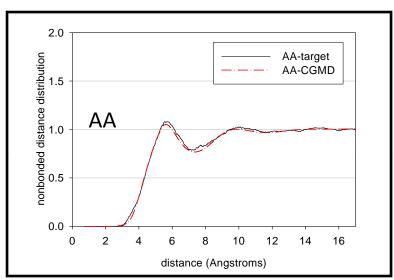
The distributions of beads with bonded interactions from MD simulations match relatively well.

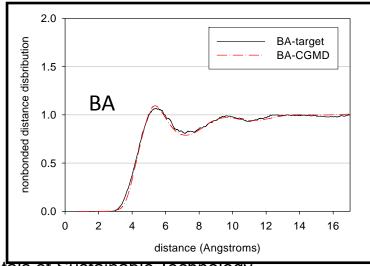
Coarse-Grained MD simulation: validation



Comparison of non-bonded PCFs (BB, BA and AA) from CGMD and atomistic MD simulations.







The non-bonded PCFs from CGMD and atomistic MD simulations match fairly well.

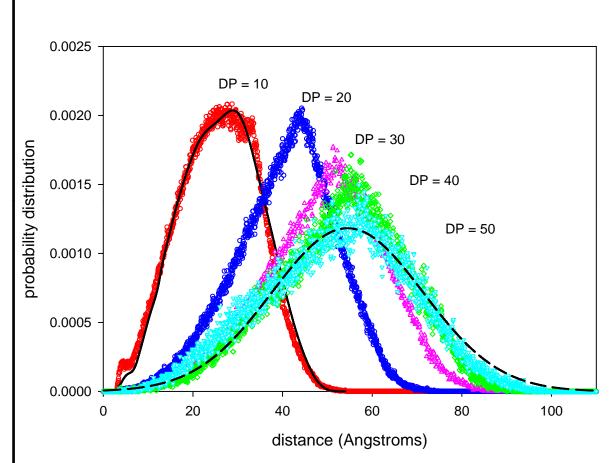


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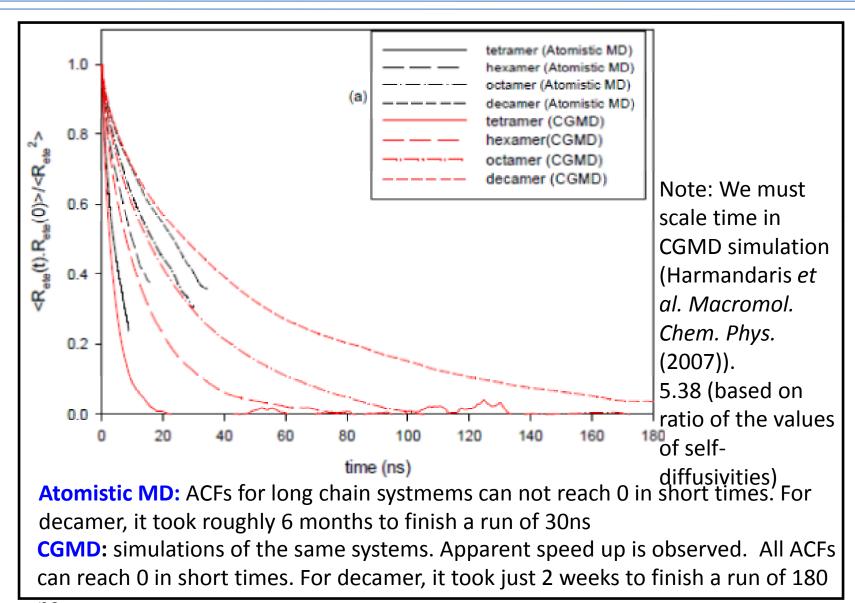
Decamer:

Comparison of chain end-to-end distance distributions from CGMD and atomistic MD match fairly well.

➤ The distributions are not typical Gaussian distributions, with chain length increases, they become more Gaussian like.

CGMD simulation: comparison of dynamic property chain end-to-end distance auto-correlation function (ACF)







Scaling exponents: *b* from polymer physics.

$$X = a(DP)^b$$

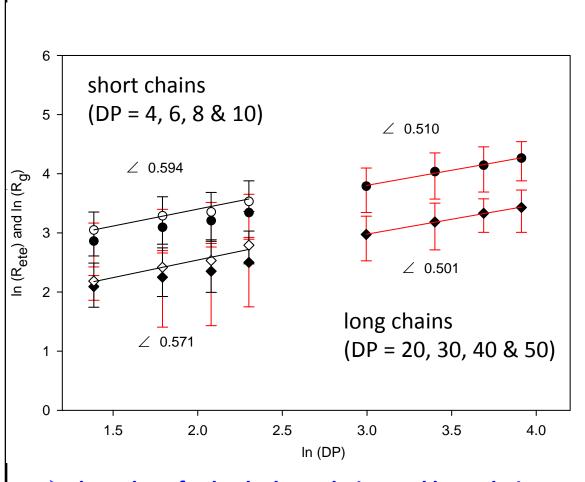
DP	Simulation method	D	η	τ _{KWW}	<r<sub>ete></r<sub>	<r<sub>g></r<sub>
1~10	Atomistic MD	-2.01	0.96	2.78	0.594	0.571
4~10	CGMD	-1.91	1.6	2.81	0.59	0.57
20~50	CGMD	-2.0	2.0	3.7	0.51	0.50
Rouse Model (1, 2)	N/A	-1	1	2	0.59	0.59
Reptation Model (1, 2)	N/A	-2	3	3	0.50	0.50

¹ Tzoumanekas et al. Macromolecules 2009

² Lahmar et al. Macromolecules 2009

Coarse-Grained MD simulation: Scaling exponents of R_{ete} and R_g



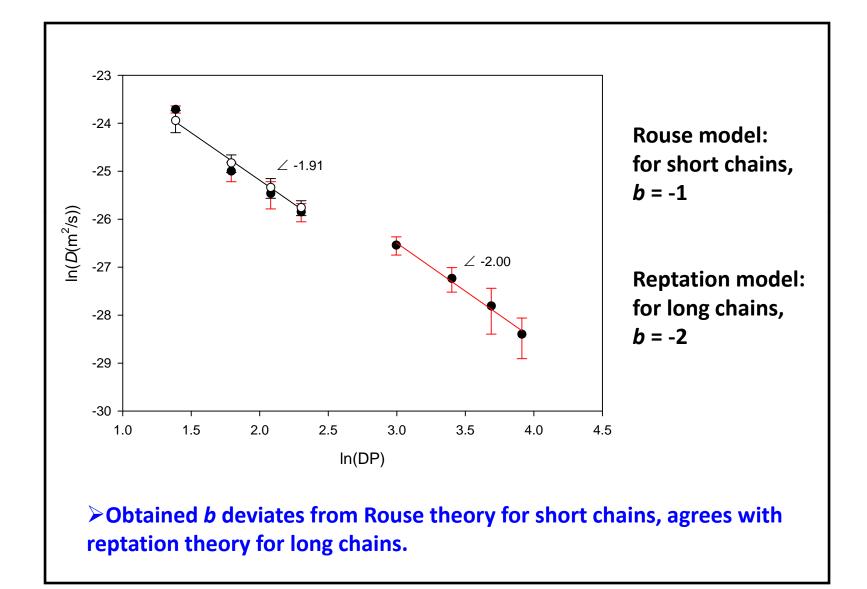


for Long chains: obtained *b* is close to 0.5

for short chains: obtained b is in the range of 0.57~0.60

➤ The values for both short chains and long chains agree with the Rouse and Reptation theory respectively.







To further understand the reptation behavior, we can do entanglement analysis by extracting entanglement information directly from configurations of the chains.

The Z-code: A common algorithm to study the entanglements in polymeric systems (Kroger, M. Comput. Phys. Commun. 2005)

What can we get from Z-code? mean contour length of primitive path (<Lpp>), tube diameter (d), number of monomers between entanglement points (N_e), interentanglement strand length (N_{FS}), defined as:

$$N_{ES} = \frac{N(N-1)}{Z(N-1)+N}$$

Kamio *et al*. Macromolecules **2007**, 40, 710.

Coarse-Grained MD simulation: entanglement analysis



DP	<l<sub>pp> (Å)</l<sub>	d (Å)	N _e	Z	N _{ES}
10	31.99	19.07	8.82	1.87	7.19
20	62.08	33.08	14.60	2.44	11.16
30	92.17	35.38	18.67	3.51	13.48
40	110.22	38.49	22.16	4.34	15.13
50	133.23	34.74	22.60	6.02	14.37
rheology models	N/A	35ª, 38-43 ^b	30.2 ^a , 24.2 ^b , 25.0 ^c	N/A	14.9 ^d

For DP = 10, unentangled system. for DP = 20 to 50, (d), (Ne) and (N_{ES}) are very close to the reported values for entangled PET melts.

a Fetters et al. In Physical Properties of Polymers Handbook; James E. Mark, 2007

b Fetters et al. Macromolecules 1994, 27, 4639.

c Lorentz, G.; Tassin, J. F. Polymer 1994, 35, 3200.

d Kamio et al. Macromolecules 2007, 40, 710.

Conclusions:



- In this example, a multiscale modeling approach was used to model PET.
- Atomistic MD simulations were used to model short chains, up to DP=10. From these simulations, PCFs of the proposed CG beads were generated.
- ➤ A technique (Iterative Boltzmann Inversion, or in this case the Ornstein-Zernike equation) was used to generate interaction potential for the CG beads.
- > CG dynamic simulations of short chains were performed. From these simulations, PCFs of the CG system were generated.
- ➤ The PCFs from the atomistic and CG simulations of short chains were compared to validate the CG potentials.
- ➤ Once validated, the CG potentials were used to simulate long chains, which could be compared to reptation theory and entanglement models.

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Brock Thomas (research scientist, Eastman Chemical Company)

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UT Computational Materials Research Group